Fathy Mohamed Abel Aziz El-Taweel* [a] and Mohamed Hilmy Elnagdi [b]
[a] Department of Chemistry ; Faculty of Science; New Damietta, University of Mansoura, A. R. Egypt
[b] Department of Chemistry; Faculty of Science; Cairo University, Giza- A. R. Egypt
Received February 6, 2001


#### Abstract

3-Acetylcoumarine was condensed with dimethylformamide dimethylacetal (DMFDMA) to yield the enaminone, which reacts readily with hydroxylamine and with hydrazines to yield coumarin-3-ylisoxazoles and coumarin-3-ylpyrazoles respectively. Reaction of the enaminone with benzamidine hydrochloride and 3 -amino-1,2,4-1 H -triazole affords the pyrimidine and triazolo[3,4-b]pyrimidine. The enaminone reacts with hippuric acid and with the dithiocarboxylic acid to yield pyranones. The reaction of the enaminone with 3 -amino-1H-1,2,4-triazole gives the triazolo[3,4-b]pyrimidine. The enaminone underwent self dimerization on reflux in acetic acid ammonium acetate to yield the coumarinyl pyridines and reacted with ketone under the same conditions to yield the pyridine. The reaction of the enaminone with 1,4-benzoquinone and 1,4 -naphthoquinone gives benzofuryl coumarine derivatives.


J. Heterocyclic Chem., 38, 981 (2001).

Enaminones are versatile reagents and their utility in heterocyclic synthesis has recently received a considerable attention [1-4]. In conjunction with our interest in synthesis of 3-heteroarylcoumarine derivatives as potential laser dyes [5], pharmaceuticals [6] and agrochemicals [7]; we report here an efficient synthesis of a variety of target molecules utilizing the readily obtainable 3 -acetylcoumarine $\mathbf{1}$ as starting material. Thus, 3-acetylcoumarine 1 reacted with dimethylformamide dimethylacetal (DMFDMA) in refluxing xylene to yield the enaminone 2. The ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $\mathbf{2}$ shows that the trans isomer is formed exclusively based on coupled olefinic protons at $\delta 5.95$ and $\delta=8.5 \mathrm{ppm}$ with $J=15 \mathrm{~Hz}$ (AB system). The enaminone 2 reacted with hydroxylamine hydrochloride in the presence sodium acetate in EtOH to yield a product that may be formulated as the isoxazolylcoumarine $\mathbf{3}$ or $\mathbf{4}$. Structure $\mathbf{3}$ was established as the product because it is identical with product obtained from the reaction of oxime 5 with DMFDMA in refluxing xylene. It is thus assumed that compound $\mathbf{3}$ is formed via initial addition of the hydroxylamine nitrogen with the carbonyl group in system $\mathbf{2}$ and the resulting product then cyclizes into the final isoxazole derivatives 3 (Figure 1).


Similarly, compound 2 reacted with hydrazine hydrate and with phenylhydrazine to yield the pyrazole derivatives 6. Attempts to obtain the pyrazole 7 from the reaction of the hydrazone $\mathbf{8}$ with DMFDMA failed. The enaminone 2 also reacted with benzamidine hydrochloride 9 to yield the 4-coumarinoyl pyrimidine 10 (Figure 2).


Figure 2

3-Amino-1 H -1,2,4-triazole 11 reacted with 2 to yield a 1,2,4-triazolopyrimidine derivative that may be formulated as $\mathbf{1 2}$ or isomeric 13. Structure $\mathbf{1 2}$ was established as the reaction product based on the ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum, where the triazole proton is observed at $\delta 8.8 \mathrm{ppm}$, which is



19





21


15
$\downarrow 2$



17

Figure 3
of 2 yielding 16 followed by further rearrange to give 17. This is similar to the well known Kepe [8] pyranone synthesis that has recently been adopted by Elnagdi et al. [6] to enable synthesis of pyranones from reaction of the enaminone 2 with hippuric acid. Compound 2 also reacted with the dithiocarboxylic acid $\mathbf{1 8}$ to yield a product that may formulated as $\mathbf{2 0}$ and 21. Formation of this product may be assumed to proceed via initial cyclization of the dithiocarboxylic acid derivatives into the thiazolone 19. This then condensed with 2 via dimethylamine elimination to yield 20 that rearranges into a pyranone derivative 21 in a manner similar to that suggested to account for formation of pyranones from reaction of hippuric acid with 2 (Figure 3).

Refluxing of 2 in acetic acid/ammonium acetate afforded the pyridine derivative 23 (Figure 4). Compound $\mathbf{2}$ also reacted with $\mathbf{1}$ in acetic acid/ammonium acetate to yield the pyridine derivative $\mathbf{2 4}$. Compound 2 reacted with 1,4-benzoquinone 25 and 1,4-naphthoquinonone 27 yielding the benzofuran 26 and naphthofuran 28, espectively. It is believed that $\mathbf{2}$ first add the quinone to yield intermediate phenolic adduct that cyclizes via loss of dimethylamine to yield final isolable products. Similar reaction sequence has been recently proposed to account for formation of benzofuran from reaction of enaminones with quinones [4] (Figure 4).


Figure 4
shifted from the expected value for the triazole proton of isomeric $\mathbf{1 3}$ by 1.0 ppm . This effect results from Van der Waals deshielding of the coumarinyl $\mathrm{H}-4$, which in turn is deshielded and appears as singlet at $\delta 8.0 \mathrm{ppm}$.

Reacting 2 with hippuric acid 14 in acetic anhydride afforded the 3-pyranylcoumarine derivative 17. These are assumed to be formed via initial cyclization of hippuric acid into the oxazolone 15 which then adds to the activated double bond system of another molecule

EXPERIMENTAL

All melting points are uncorrected. IR spectra were recorded for KBr disks on a Shimadzu IR-740 spectrometer. ${ }^{1} \mathrm{H}$ NMR spectra were obtained on a Bruker AC-80 spectrometer with DMSO- $\mathrm{d}_{6}$ as solvent and TMS as internal standards and chemical shifts expressed $\delta \mathrm{ppm}$. Mass spectra were measured on GC-MS INCOS XL Finnigan MAT. Elemental analyses were performed on LECO CHNS-932.

1-(3-Coumarinyl)-3-dimethylamino-2-propen-1-one (2).
Dimethylformamide dimethylacetal ( $1.19 \mathrm{~g}, 10 \mathrm{mmol}$ ) was added to 3-acetylcoumarin $\mathbf{1}(10 \mathrm{mmol})$ in xylene ( 50 ml ), and the reaction mixture was refluxed for 6 hours. The removal of solvent under reduced pressure yielded the crude product which was crystallized from ethanol. Compound 2 was obtained as yellow crystals ( $75 \%$ ), mp $165{ }^{\circ} \mathrm{C}$; ir (potassium bromide): $v$ 1720 (ring CO) $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (dimethyl-d ${ }_{6}$ sulfoxide): $\delta 2.90$ (s, $3 \mathrm{H}, \mathrm{NCH}_{3}$ ), $3.15\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 5.95(\mathrm{~d}, J=15 \mathrm{~Hz}$, olefinic-H), 7.35-7.90 (m, 5H, arom-H), $8.50(\mathrm{~d}, J=15 \mathrm{~Hz}$, olefinic-H); ms : 243 ( $\mathrm{M}^{+}$).

Anal. Calcd. for $\mathrm{C}_{14} \mathrm{H}_{13} \mathrm{NO}_{3}: \mathrm{C}, 69.13 ; \mathrm{H}, 5.39 ; \mathrm{N}, 5.76$. Found: C, 69.22; H, 5.34; N, 5.66.

3-(Coumarin-3'-yl)isoxazole (3).
A mixture of the enaminone $\mathbf{2}(2.43 \mathrm{~g}, 10 \mathrm{mmol})$ and hydroxylamine hydrochloride ( 10 mmol ) in ethanol ( 50 ml ), sodium acetate anhydrous ( $0.82 \mathrm{~g}, 10 \mathrm{mmol}$ ) was refluxed for 6 hours, then left to cool. The formed precipitate was collected by filtration and crystallized from ethanol. Compound $\mathbf{3}$ was obtained as colorless crystals ( $65 \%$ ), mp $195^{\circ} \mathrm{C}$; ir (potassium bromide): $v$ 1726 (ring CO) $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$-NMR (dimethyl- $\mathrm{d}_{6}$-sulfoxide): $\delta 7.25$ (d, $J=3 \mathrm{~Hz}, 1 \mathrm{H}$, isoxazoleH-4), $8.75(\mathrm{~d}, J=3 \mathrm{~Hz}, 1 \mathrm{H}$, isoxazole H-5) 7.41-8.0 (m, 4H, Ar-H), 8.9 (s, 1H, coumarin H-4); ms: 213 $\left(\mathrm{M}^{+}\right)$.
Anal. Calcd. for $\mathrm{C}_{12} \mathrm{H}_{7} \mathrm{NO}_{3}: \mathrm{C}, 67.61 ; \mathrm{H}, 3.31 ; \mathrm{N}, 6.57$. Found: C, 67.50; H, 3.36; N, 6.42.
Reaction of Compound 2 with Hydrazines.
To a solution of $\mathbf{2}(2.43 \mathrm{~g}, 10 \mathrm{mmol})$ in ethanol ( 30 ml ), either hydrazine hydrate ( 2 ml ) or phenylhydrazine $(1.5 \mathrm{ml}$ ) were added. The reaction mixture was heated under reflux for 4 hours, then left to cool. The solid products, so formed, were isolated by filtration, washed with ethanol and dried. Recrystallization from ethanol yielded colorless crystals of $\mathbf{6 a , b}$.

## 3-(Coumarin-3'-yl)pyrazole (6a).

Compound 6a was obtained in $60 \%$ yield $\mathrm{mp} 237^{\circ} \mathrm{C}$; ir (potassium bromide): v $3200(\mathrm{NH}), 1736$ (ring CO), $1626(\mathrm{C}=\mathrm{N}) \mathrm{cm}^{-1}$, ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (dimethyl-d ${ }_{6}$ sulfoxide): $\delta 5.9$ (d, $J=3 \mathrm{~Hz}, 1 \mathrm{H}$, pyrazole H-4), 7.35-7.75 (m, 4H, Ar-H), 8.45 (d, $J=3 \mathrm{~Hz}, 1 \mathrm{H}$, pyrazole H-5), $9.0(\mathrm{~s}, 1 \mathrm{H}$, coumarin $\mathrm{H}-4), 11.1(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH})$; ms: 212 $\left(\mathrm{M}^{+}\right)$.
Anal. Calcd. for $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 67.92; H, 3.80; N, 13.20. Found: C, 67.78; H, 3.5; N, 13.11.

## 1-Phenyl-3-(coumarin-3'-yl)pyrazole (6b).

Compound 6b was obtained in $60 \%$ yield, mp $135{ }^{\circ} \mathrm{C}$; ir (potassium bromide): $v 1736$ (ring CO), $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}-\mathrm{NMR}$ (dimethyl-d $\mathrm{d}_{6}$ sulfoxide): $\delta 6.8(\mathrm{~d}, J=3 \mathrm{~Hz}, 1 \mathrm{H}$, pyrazole $\mathrm{H}-4$ ), 7.35-7.83 (m, 4H, Ar-H), 7.84 (d, $J=3 \mathrm{~Hz}, 1 \mathrm{H}$, pyrazole $\mathrm{H}-5$ ), $8.21\left(\mathrm{~s}, 1 \mathrm{H}\right.$, coumarin H-4); ms: $2289\left(\mathrm{M}^{+}\right)$.

Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 74.99; H, 4.20; N, 9.72. Found: C, 74.89; H, 4.12; N, 9.69.
4-(Coumarin-3'-yl)-2-phenylpyrimidine (10).
A solution of $2(2.43 \mathrm{~g}, 10 \mathrm{mmol})$ in dry pyridine $(20 \mathrm{ml})$, and $(1.55 \mathrm{~g}, 10 \mathrm{mmol})$ of benzamidine hydrochloride were refluxed for 3 hours. The solvent was reduced under vacuo, to half volume. The solid product, obtained, was isolated by filtration, dried
and recrystallized from ethanol. Compound $\mathbf{1 0}$ was obtained as buff crystals, yield $70 \%, \mathrm{mp} 185{ }^{\circ} \mathrm{C}$; ir (potassium bromide): v 1732 (ring CO), $1606(\mathrm{C}=\mathrm{N}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}-\mathrm{NMR}$ (dimethyl $-\mathrm{d}_{6}$ sulfoxide): $\delta 7.39-7.90(\mathrm{~m}, 9 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 8.20(\mathrm{~s}, 1 \mathrm{H}$, coumarin H-4), 8.96 (d, $J=3 \mathrm{~Hz}, 1 \mathrm{H}$, pyridine H-6), 9.03 (d, $J=3 \mathrm{~Hz}, 1 \mathrm{H}$, pyrdine $\mathrm{H}-5)$; ms: $300\left(\mathrm{M}^{+}\right)$.

Anal. Calcd. for $\mathrm{C}_{19} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 75.99; H, 4.03; $\mathrm{N}, 9.33$. Found: C, 75.5; H, 4.60; N, 9.41.
4-(Coumarin-3'-yl)-1,2,4-triazolo[4,3-a]pyrimidine (12).
A solution of $2(2.43 \mathrm{~g}, 10 \mathrm{mmol})$ and $(0.84 \mathrm{~g}, 10 \mathrm{mmol})$ of 3 -amino- 1 H -1,2,4-triazole in ( 30 ml ) ethanol/acetic acid (1:1), was refluxed for 6 hours. The solvent was removed by distillation under reduced pressure and the resulting solution was left to cool. The solid precipitate was collected by filtration and recrystallized from ethanol. Compound $\mathbf{1 2}$ was obtained as canary yellow crystals, yield $70 \%, \mathrm{mp} 250^{\circ} \mathrm{C}$; ir (potassium bromide): v 1729 (ring CO), $1653(\mathrm{C}=\mathrm{N}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (dimethyl- $\mathrm{d}_{6}$ sulfoxide): $\delta$ 7.35-7.50 (m, 2 H , arom-H), 7.60-7.73 (m, 2 H , arom-H), 7.8-7.9 (d, $J=\mathrm{Hz}$, pyrimidine $\mathrm{H}-5$ ), 8.5 ( $\mathrm{s}, 1 \mathrm{H}$, coumarinyl $\mathrm{H}-4$ ), 8.8 ( $\mathrm{s}, 1 \mathrm{H}$, triazole $\mathrm{H}-3$ ), 9.1 ( $\mathrm{s}, 1 \mathrm{H}$, pyrimidine $\mathrm{H}-6$ ); ms: $264\left(\mathrm{M}^{+}\right)$.

Anal. Calcd. for $\mathrm{C}_{14} \mathrm{H}_{8} \mathrm{~N}_{4} \mathrm{O}_{2}$ : C, 63.64; H, 3.05; N, 21.20. Found: C, 63.41; H, 4.60; N, 21.35.

General Procedure for the Preparation of 3-Substituted-6-(coumarin-3'-yl)pyran-2-one 17 and 21.

A solution of $2(2.43 \mathrm{~g}, 10 \mathrm{mmol})$ and appropriate amount of $\mathbf{1 4}$ or $\mathbf{1 8}(10 \mathrm{mmol})$ in acetic anhydride ( 50 ml ), was refluxed for 3 hours, then left to cool. The deposited solids were isolated by filtration and recrystallized to give 17 and 21, respectively.

## 3-Benzamido-6-(coumarin-3'-yl)pyran-2-one (17).

Compound $\mathbf{1 7}$ was obtained as canary yellow crystals from DMF yield $75 \%$, mp $284^{\circ} \mathrm{C}$; ir (potassium bromide): v 3396 (NH), 1726 (ring CO), 1695 (ring CO), 1664 (amide CO) cm-1; ${ }^{1} \mathrm{H}$ NMR (dimethyl-d $\mathrm{d}_{6}$ sulfoxide): $\delta 7.35-7.50(\mathrm{~m}, 2 \mathrm{H}$, arom-H), $7.3-7.6(\mathrm{~m}, 7 \mathrm{H}$, arom-H), $7.85(\mathrm{~d}, J=8 \mathrm{~Hz}, 1 \mathrm{H}$, pyrane $\mathrm{H}-5), 8.30$ (d, $J=8 \mathrm{~Hz}, 1 \mathrm{~h}$, pyrane H-4), 8.75 ( $\mathrm{s}, 1 \mathrm{H}$, coumarinyl H-4), 9.74 (s, 1H, NH); ms: $359\left(\mathrm{M}^{+}\right)$.

Anal. Calcd. for $\mathrm{C}_{21} \mathrm{H}_{13} \mathrm{NO}_{5}$ : C, $70.19 ; \mathrm{H}, 3.65$; N, 3.90. Found: C, 70.31; H, 4.10; N, 4.02.

3-Benzylthiocarboxamido-6-(coumarin-3'-yl)pyran-2-one (21).
Compound 21 was obtained as buff crystals from ethanol/DMF yield $70 \%, \mathrm{mp} 225{ }^{\circ} \mathrm{C}$; ir (potassium bromide): $v$ 3450 (NH), 1746 ( CO), 1720 (ring CO) cm ${ }^{-1}$; ${ }^{1} \mathrm{H}$ NMR (dimethyl-d ${ }_{6}$ sulfoxide): $\delta 2.84\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 6.78(\mathrm{~d}, J=8 \mathrm{~Hz}$, 1 H , pyrone H-4), 7.25-7.59 (m, 9H, Ar-H), $7.89(\mathrm{~d}, J=8 \mathrm{~Hz}, 1 \mathrm{H}$, pyrone H-5), 8.30 ( $\mathrm{s}, 1 \mathrm{H}$, coumarinyl $\mathrm{H}-4$ ), 8.34 ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{NH}$ ); ms: $421\left(\mathrm{M}^{+}\right)$.

Anal. Calcd. for $\mathrm{C}_{22} \mathrm{H}_{15} \mathrm{NO}_{4} \mathrm{~S}_{2}$ : C, 62.69; H, 3.59; N, 3.32. Found: C, 62.63; H, 3.41; N, 3.22.

## 2-(Coumarin-3'-yl)-5-(coumarin-3'-oyl)pyridine (23).

Compound $2(2.43 \mathrm{~g}, 10 \mathrm{mmol})$ and $(0.77 \mathrm{~g}, 10 \mathrm{mmol})$ of ammonium acetate were refluxed in glacial acetic acid ( 30 ml ) for 0.5 hour, then left to cool to room temperature. The precipitated material upon cooling was isolated by filtration and recrystallized from DMF to give 23 as buff crystals yield $65 \%$, mp $265{ }^{\circ} \mathrm{C}$; ir (potassium bromide): $v 1730$ (ring CO), 1680 (CO) $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (dimethyl-d ${ }_{6}$ sulfoxide): $\delta 7.35-7.72(\mathrm{~m}, 8 \mathrm{H}$,

Ar-H), $7.89(\mathrm{~d}, J=8 \mathrm{~Hz}, 1 \mathrm{H}$, pyridine $\mathrm{H}-3), 8.20(\mathrm{~d}, J=8 \mathrm{~Hz}, 1 \mathrm{H}$, pyridine H-4), $8.50(\mathrm{~d}, J=8 \mathrm{~Hz}, 1 \mathrm{H}$, pyridine $\mathrm{H}-5), 8.9,9.0$ $\left(2 \mathrm{~s}, 2 \mathrm{H}\right.$, coumarinyl H-4), $9.34(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH})$; ms: $395\left(\mathrm{M}^{+}\right)$.

Anal. Calcd. for $\mathrm{C}_{24} \mathrm{H}_{13} \mathrm{NO}_{5}: \mathrm{C}, 72.91 ; \mathrm{H}, 3.31 ; \mathrm{N}, 3.54$. Found: C, 72.85; H, 3.24; N, 3.46.

2,6-Bis(coumarin-3'-yl)pyridine (24).
A mixture of $2(2.43 \mathrm{~g}, 10 \mathrm{mmol})$ and $(1.88 \mathrm{~g}, 10 \mathrm{mmol})$ of 3 -acetylcoumarin 1 in glacial acetic acid $(30 \mathrm{ml})$ and $(0.77 \mathrm{~g}, 10$ mmol ) of ammonium acetate, was refluxed for 1 hour. The precipitated material was isolated by filtration and recrystallized. Compound 24 was obtained as yellow crystals from ethanol/DMF yield $65 \%, \operatorname{mp} 280^{\circ} \mathrm{C}$; ir (potassium bromide): $v$ 1730 (ring CO), 1673 (CO) $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (dimethyl-d ${ }_{6}$ sulfoxide): $\delta 7.6-8.4(\mathrm{~m}, 11 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 9.03,9.06(2 \mathrm{~s}, 2 \mathrm{H}$, coumarinyl H-4); ms: $367\left(\mathrm{M}^{+}\right)$.

Anal. Calcd. for $\mathrm{C}_{23} \mathrm{H}_{13} \mathrm{NO}_{4}: \mathrm{C}, 75.20 ; \mathrm{H}, 3.57 ; \mathrm{N}, 3.81$. Found: C, 75.33; H, 3.61; N, 3.49.

## General Method for Preparation of 26 and 28.

To a stirred solution of $2(2.43 \mathrm{~g}, 10 \mathrm{mmol})$ in glacial acetic acid ( 50 ml ), each of $p$-benzoquinone and 1,4-naphthoquinone ( 10 mmol ) was added. Stirring was continued over night at room temperature. The reaction mixture was evaporated in vacuo and the solid products obtained were isolated by filtration and recrystallized.

## 5-Hydroxy-3-(coumarin-3'-oyl)benzo[b]furan (26).

Compound 26 was obtained as colorless crystals from ethanol/DMF yield $70 \%$, mp $284{ }^{\circ} \mathrm{C}$; ir (potassium bromide): $v$ 3337 ( OH), 1714 (ring CO), 1690 (CO) $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (dimethyl-d ${ }_{6}$ sulfoxide): $\delta 7.35-7.98(\mathrm{~m}, 7 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 8.47(\mathrm{~s}, 1 \mathrm{H}$, $\mathrm{OH})$; ms: $306\left(\mathrm{M}^{+}\right)$.

Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{10} \mathrm{O}_{5}: \mathrm{C}, 70.59 ; \mathrm{H}, 3.29$. Found: C , 70.31; H, 3.46.

5-Hydroxy-3-(coumarin-3'-oyl)naphtho[1,2-b]furan (28).
Compound 28 was obtained as yellow crystals from DMF yield $60 \%, \operatorname{mp} 280^{\circ} \mathrm{C}$; ir (potassium bromide): $v 1725$ (ring CO), $1640(\mathrm{CO}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (dimethyl-d 6 sulfoxide): $\delta 7.34-8.29$ ( $\mathrm{m}, 9 \mathrm{H}, \mathrm{Ar}-\mathrm{H}$ ), $8.50(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}), 9.0(\mathrm{~s}, \mathrm{H}$, coumarinyl H-4), 10.4 ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{CHO}$ ).

Anal. Calcd. for $\mathrm{C}_{22} \mathrm{H}_{12} \mathrm{O}_{5}$ : C, 74.16; $\mathrm{H}, 3.39$. Found: C , 74.23; H, 3.51.

## REFERENCES AND NOTES

[1] F. Al-Omran, M. M. Abdel-Khalik, A. Abuel-Khair, and M. H. Elnagdi, Synthesis, 91, (1997).
[2] K. M. Al-Zaydi and E. A. Hafez, J. Chem. Res. (S) 360; (M) 1621 (1999).
[3] A. A. Al-Naggar; M. M. Abdel-Khalik and M. H. Elnagdi J. Chem. Research (S), 648, (M), 2801, (1999).
[4] B. Al-Saleh, N. A. Al-Awedi, H. Al-Kandari, M. M. AbdelKhalik, and M. H. Elnagdi, J. Chem. Research (S), 16; (M), 201, (2000).
[5] M. H. Elnagdi, S. O. Abdallah, K. M. Ghoneim, E. M. Ebeid and K. N. Kassab, J. Chem. Research (S), 44 (1997).
[6] F. Al-Omran; N. A. Al-Awedi, M. M. Abdel-Khalik,K. Kaul, A. Abuel-Khair, and M. H. Elnagdi, J. Chem. Research (S), 84; (M), 601, (1997).
[7] E. A. Hafez, M. H. Elnagdi, A. A. Elagamey, F. M. A. ElTaweel, Heterocycles, 26, 903, (1987).
[8] V. Kepe, M. Kocevar and S. Polanc, J. Heterocyclic. Chem., 33, 1707, (1996).
[9] P. Fossa, R. Boggia, E. Lo Presti, L. Mosti, P. Dorigo and M. Floreani, Il. Farmaco, 52, 523, (1997).

